



#### Helmholtz-Zentrum Geesthacht

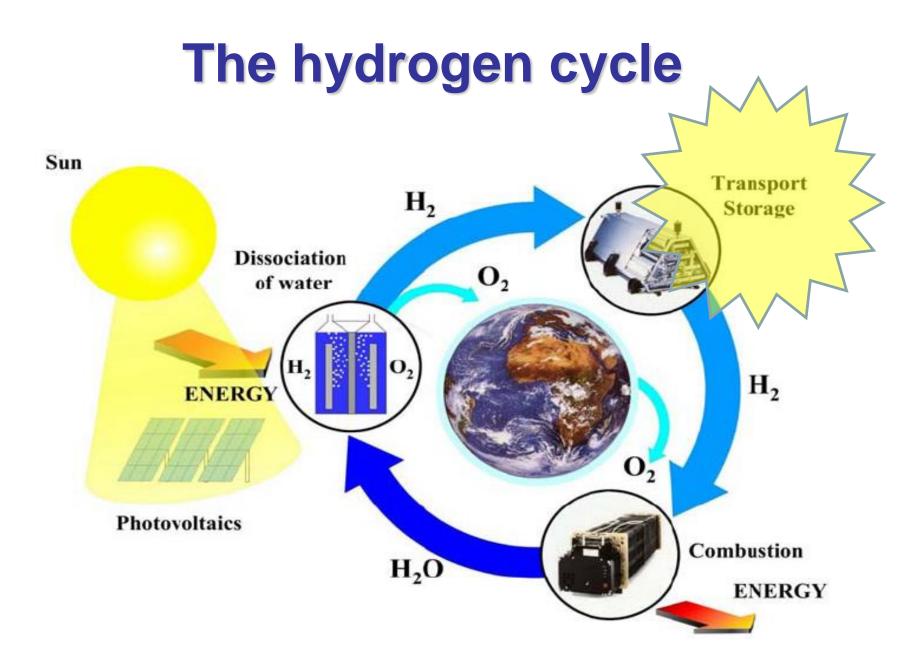
Zentrum für Material- und Küstenforschung



# Hydrogen storage and transport in a decarbonised Europe *M. Baricco<sup>1</sup>*, *K.Taube*<sup>2</sup>

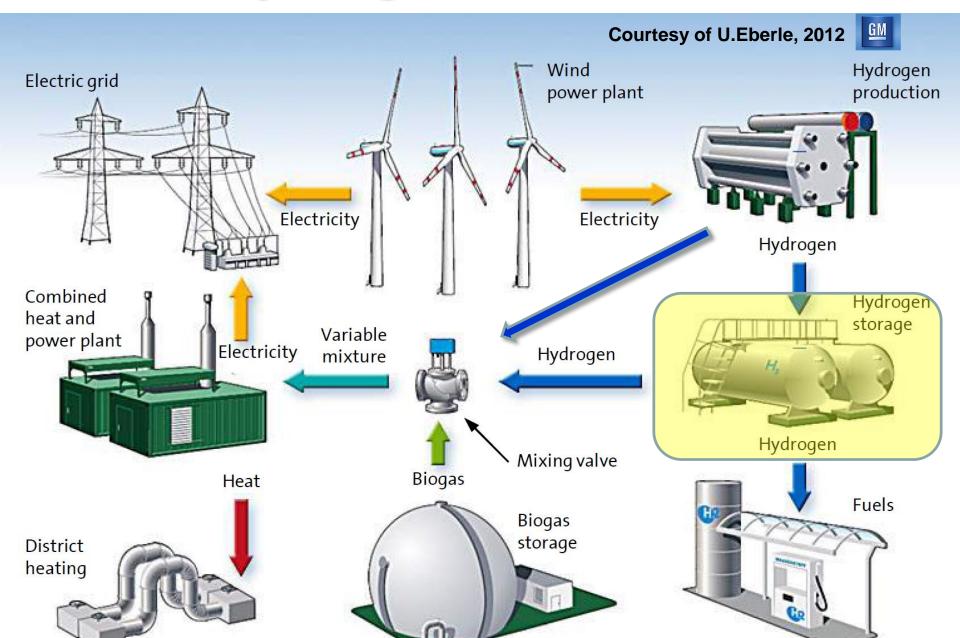
### <sup>1</sup>Department of Chemistry and NIS Via Pietro Giuria 7, 10125 Torino, Italy

<sup>2</sup>Helmholtz-Zentrum Geesthacht Zentrum für Material- und Küstenforschung GmbH Geesthacht, Germany



A.Züttel, A.Borgschulte, L. Schlapbach «Hydrogen as a Future Energy Carrier» Wiley-VCH, 2008

### **Green hydrogen from renewables**



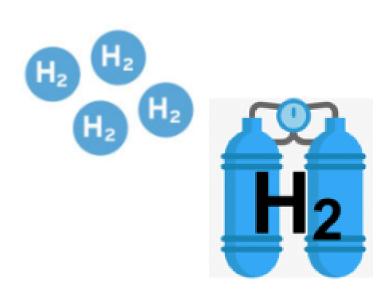
# Hydrogen handling

### Storage

Transport

Purification

Compression



Reduction of volume of gas (1 kg

 $H_2$  in N.C. about 11 m<sup>3</sup>) by:

- 1. Compression
- 2. Temperature below critical point
- 3. Reduction of repulsion by interaction with solids
- 4. Reversible reactions (no C-H covalent bonds)



SP 6 Hydrogen Production and Handling

SP 7 Hydrogen Storage





gas



# Hydrogen handling requirements



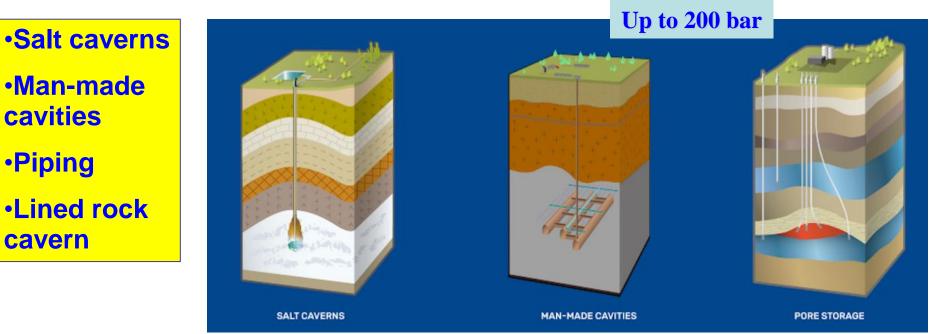
- Temperatures and pressures close to ambient conditions
- High number of charge/discharge cycles
- Fast kinetics of charge/discharge
- Safe and at low cost







### Low-pressure gas



#### •Storage at tons level

•Low compression costs

CH<sub>2</sub>ANGE from Geostock

Low construction costs
Low leakage rates
Low contaminations
Projects running

# **High-pressure gas**

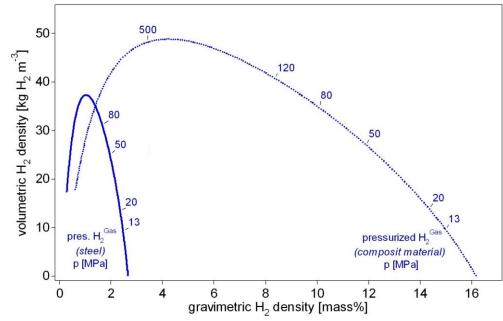
#### 700 bar!

#### Commercially available

•Storage capacity depends on the container material

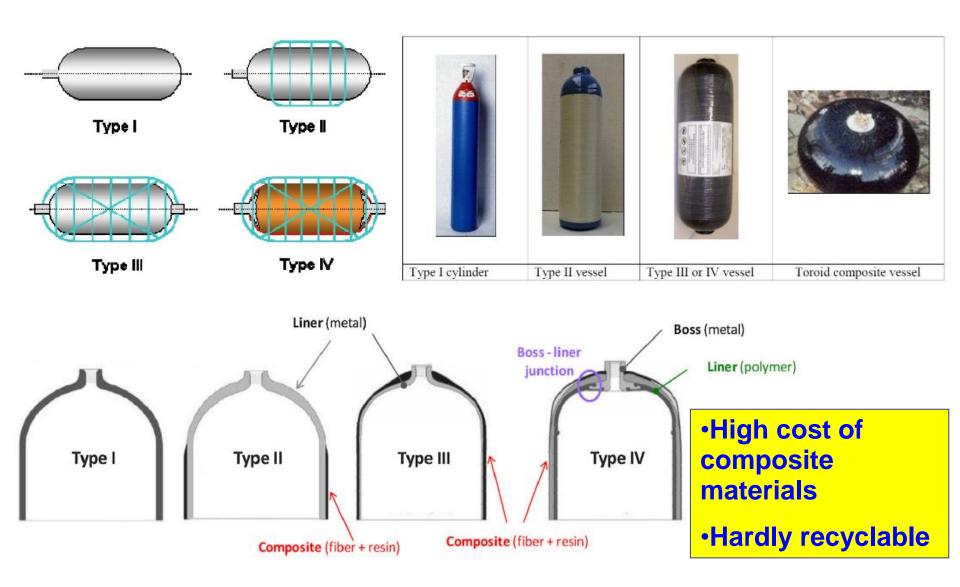


- Energy loss for compression (about 12%)
- Safety
- Pressure dropping
- Gravimetric density decreases with increasing pressure



F. Vollrath & D. P. Knight, NATURE 410 (2001), pp. 541-549

# **High-pressure gas**



# Liquid Hydrogen

Commercially available

•Storage as cryogenic liquid below the boiling point of 21.2 K

•Gravimetric density depends on the size of the container  $\rightarrow$ 100%

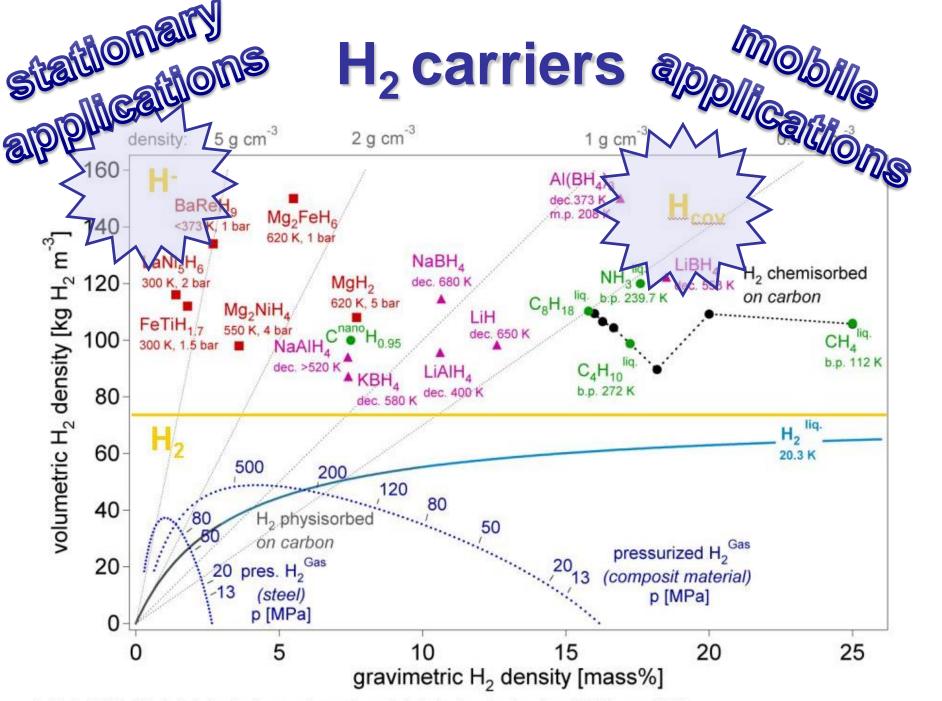
•Energy loss for liquefaction (about 30%)

•Boil-off: continuous loss of few % per day

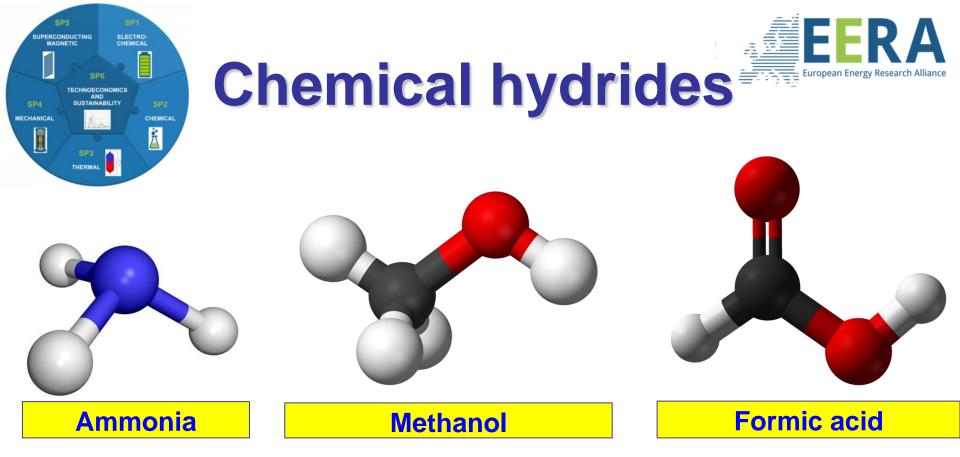
Needs of big-sized tanks

•Density ( $H_2$  liq.) = 70.8 kg·m<sup>-3</sup>





Ref: A. Züttel, "Materials for hydrogen storage", materials today, September (2003), pp. 18-27

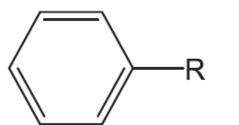


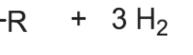
- Liquid at normal conditions
- High gravimetric density
- Catalysts are necessary
- Commercially available

### **Energy density [MJ/kg]:**

- Hydrogen 140
- Ammonia 19
- Methanol 20
- Formic acid 5

# Liquid Organic Hydrogen Carriers - LOHC

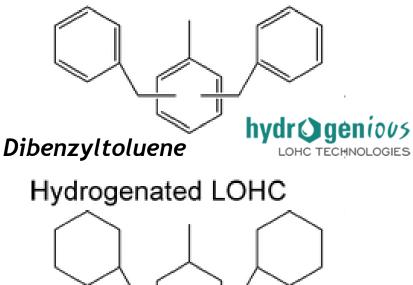






+ Heat

#### Non-Hydrogenated LOHC



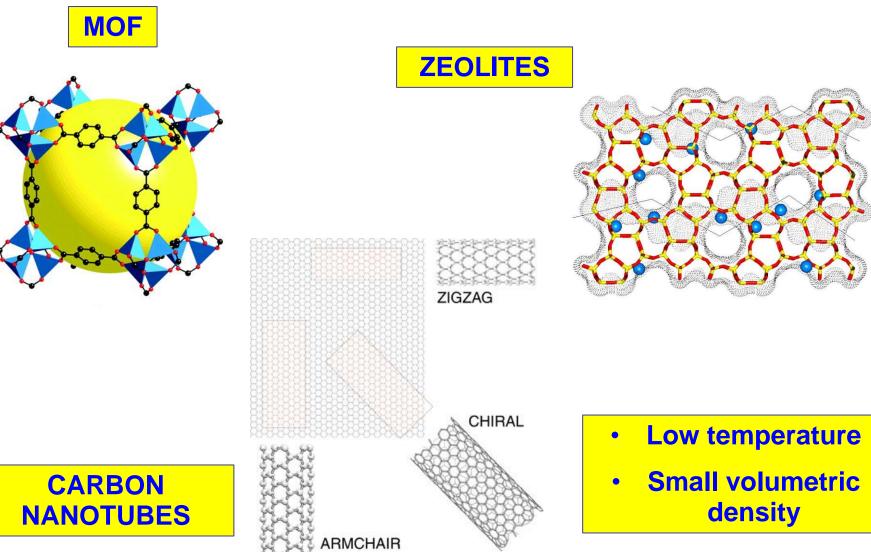
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Perhydrodibenzyltoluene

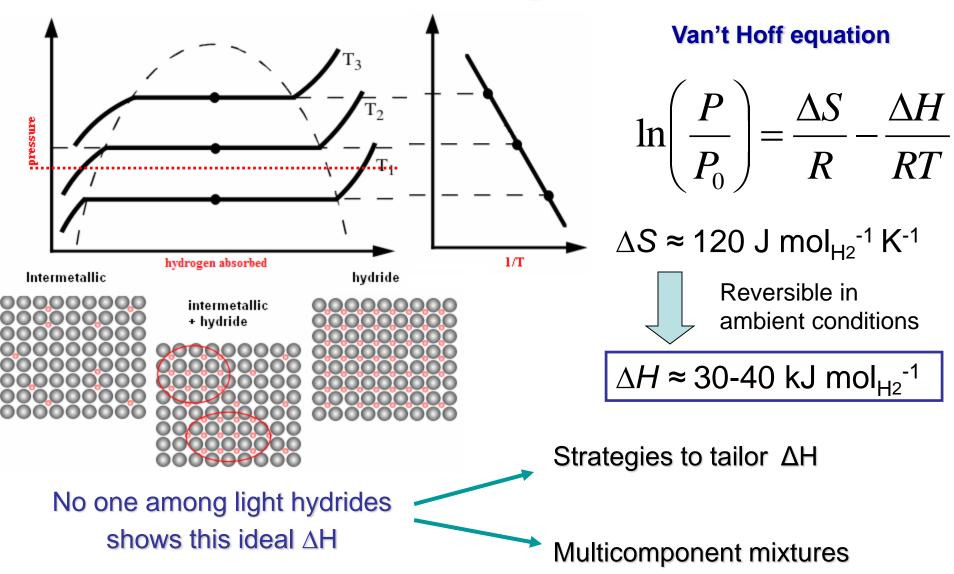
Liquid phase

- Different temperature ranges (150-350 °C)
- High gravimetric density (up to 6 wt%)
- Catalysts and purification are necessary
- Eventually 2<sup>nd</sup> store for dehydrogenated fluid necessary
- Commercially available

# Adsorption



# Hydrides for hydrogen storage





**FUEL CELLS AND HYDROGEN** JOINT UNDERTAKING



# **The HyCARE Project**



UNIVERSITÀ DEGLI STUDI DI TORINO



2019-2021





Helmholtz-Zentrum Geesthacht

Zentrum für Material- und Küstenforschung









engie

ENVIRONMENT PARK Parco Scientifico Tecnologico per l'Ambiente



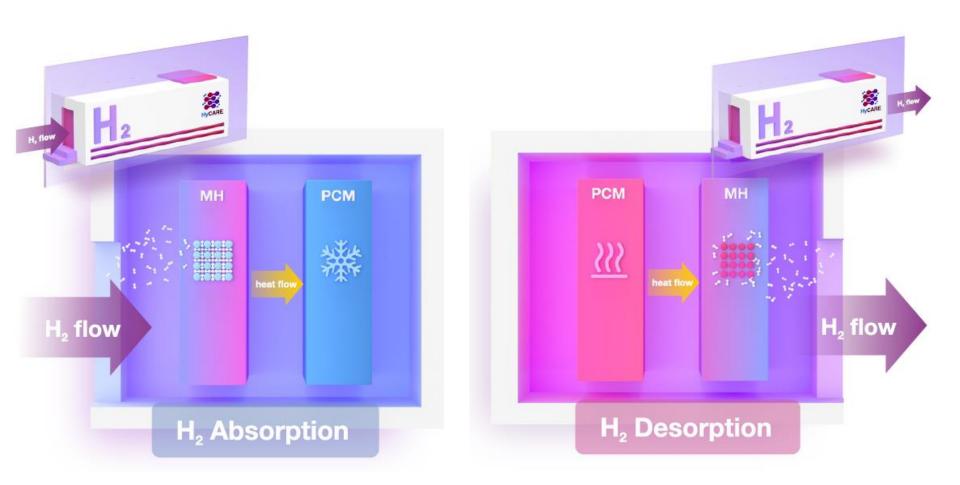
# The goals



- High quantity of stored hydrogen >= 50 kg
- Low pressure < 50 bar and low temperature < 100°C
- Low foot print, comparable to liquid hydrogen storage
- Innovative design
- Hydrogen storage coupled with thermal energy storage
- Improved energy efficiency
- Integration with an electrolyser (EL) and a fuel cell (FC)
- Demonstration in real application
- Improved safety
- Techno-economical evaluation of the innovative solution
- Analysis of the environmental impact via Life Cycle Analysis (LCA)
- Exploitation of **possible industrial applications**
- Dissemination of results at various levels
- Engagement of local people and institution in the demonstration site



### The concept





# System design & prototype test

### System design

Metal hydride

- $\approx$  4 ton of TiFe-alloy  $\approx$  40-44 kg of H<sub>2</sub>
- xx tanks
   PCM
- $\approx 2.7$  ton of PCM CRODA
- yy tanks

### Prototype design

Metal hydride module

- 60 kg of TiFe-alloy
- pellets

PCM module

• 30 kg of PCM

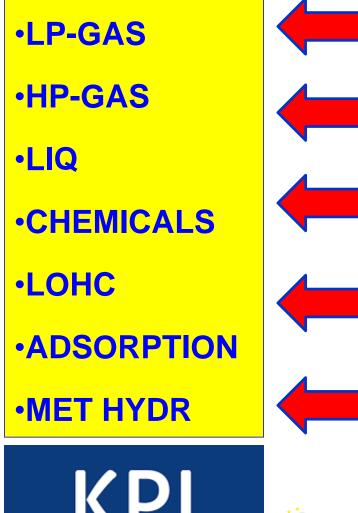
Std. ISO Container

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11		MH-TANK	
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	PCM-TANK		Ð
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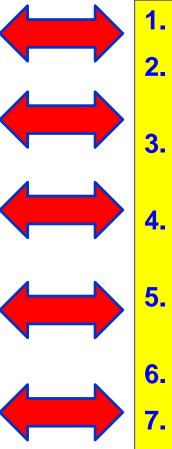
**Prototype test** 

- Models validation
- Optimization of the final design

# **Challenges for hydrogen handling**



KEY PERFORMANCE INDICATOR



- Quantity
- Gravimetric density
- Volumetric density
- Energy efficiency
- Charging time
- Reversibility
- Sensitivity to H<sub>2</sub> quality

- 9. Thermal management
- 10. Life cycle
- 11. Materials for tank
- 12. Stability on time
- 13. Availability
- 14. Safety
- 15. Cost
- 16. Acceptance

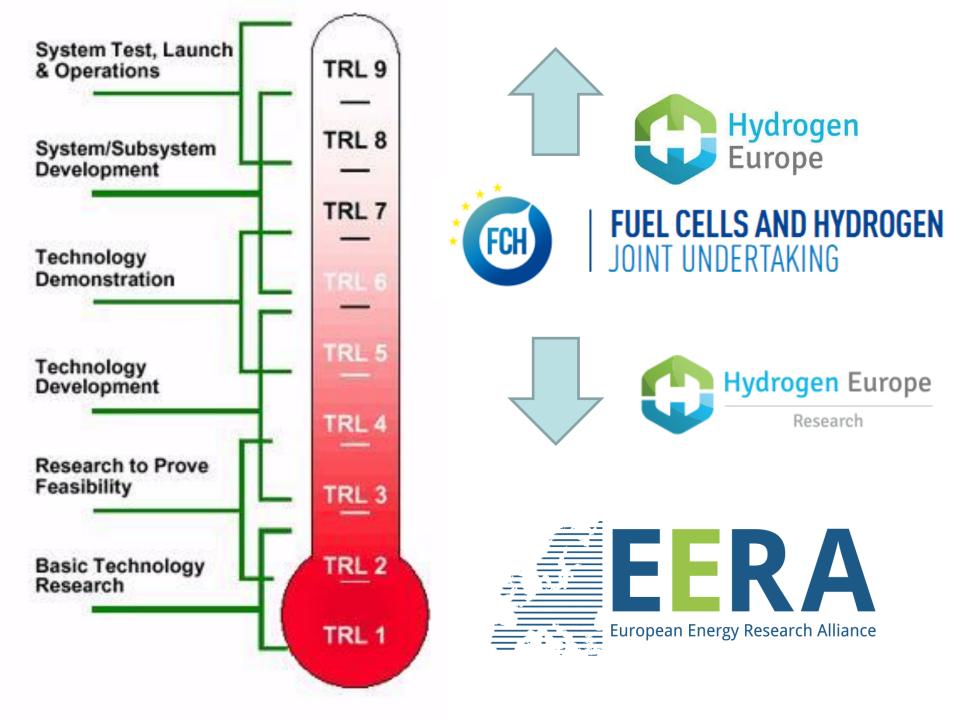




















**Specific Objective 3**: Delivering Hydrogen at low cost Roadmaps:

- large scale hydrogen storage
- hydrogen in the gas grid
- hydrogen carriers
- developing existing hydrogen transport means
- key technologies for hydrogen distribution (H<sub>2</sub> compression, metering, purification and separation)

### **Open points for the discussion**

What are optimal (energy, cost, footprint) hydrogen storage and transport options for:

- On-shore storage and transport
  - Truck
  - Train
  - In the gas grid
- Large scale off-shore power production and hydrogen generation (i.e. electrolysers directly in the wind farms)
  - Underground
  - Ship
  - Pipeline
- National, international and Intercontinental hydrogen transport for supply of Europe
  - Ship
  - Pipeline

### **Open points for the discussion**

Transport and storage options – where to use which in relation to amount of H<sub>2</sub> to be stored and transported

- Gaseous
- Liquid hydrogen
- Chemicals
  - Ammonia
  - Formic acid
  - Other
- LOHC
  - Toluene
  - Dibenzyltoluene/Marlotherm®
  - Others
- Solid carriers
  - Metal hydrides
  - Porous materials